Roll No.

(05/16-I)

5230

B. Sc. EXAMINATION

(For Batch 2013 & Onwards)

(Fourth Semester)

CHEMISTRY: ORGANIC CHEMISTRY
Paper XIII (CH-206)

Time: Three Hours Maximum Marks: 27

Note: Attempt *Five* questions in all, selecting at least *two* questions from each Section. Q. No. 1 is compulsory.

- 1. (a) For acylation reaction acetic anhydride is preferred over acetyl chloride, why? 1
 - (b) Give IUPAC nomenclature of the following organic compounds: 2

(2-17) B-5230



- Deduce the number of fundamental vibrations possible for CO₂.
- Out of trimethylamine and n-propylamine, which is having higher boiling point and why?
- What is Diazotisation? (e)
- How IR spectroscopy is helpful in differentiating between dimethyl ether and ethanol?

Section A

- Give mechanism of acidic hydrolysis of (a)
 - Explain Hell-Volhard-Zelinsky reaction. (b)
 - How will you do the following conversions ?
 - (i) Silver benzoate → Bromobenzene
 - (ii) Toluene → Benzoic acid

B-5230

- Discuss briefly the effect of substituents on the acidity of aromatic carboxylic acids.
 - (b) What is Hoffman bromide reaction? Using this reaction how will you convert benzoic acid to aniline?
 - How IR spectroscopy is helpful in determining whether a hydroxyl group in an organic compound is free or associated?
- What are fundamental vibrations? Name 4. (a) their types with examples.
 - Give selection rules for IR spectroscopy.

Give values of characteristics C=O stretching band of IR of carboxylic acid

Section B

derivatives.

Can we prepare aniline by Gabrielphthalimide reaction? Explain.

P.T.O.

	(0)	Discuss at least three methods o
		preparation of aryl amines.
6.	(a)	Give a suitable explanation for the
		following:
		(i) Aniline is weaker base than ammonia
		(ii) Ethylamine is more basic than aniline.
	(b)	What is Hoffmann bromide reaction? Explain giving mechanism.
7.	(a)	Why arenediazonium salts are highly reactive?
	(b)	What are coupling reactions? Discuss
		their synthetic applications. 4
		생님 그 경영이 되는 경영에 가장 보고 있다면 하는데 가장 하는데 되었다.